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### Stoecker Refrigeration Air Conditioning Solution

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### 317823436-Solution-Refrigeration-Air-Conditioning-Stoecker ...

3-1 Calculate the specific volume of an air-vapor mixture in cubic meters per kilogram of dry air when the following conditions prevail:  $t = 30\text{ C}$ ,  $W = 0.015\text{ kg/kg}$ , and  $p_t = 90\text{ kPa}$ . Solution: Equation 3-4.  $t_s = 30\text{ C} + 273 = 303\text{ K}$   $R_a = 287\text{ J/kg.K}$   $p_t = 90\text{ kPa} = 90,000\text{ Pa}$  Equation 3-2  $t_s = 30\text{ C} + 273 = 303\text{ K}$   $W = 0.015$   $p_t = 90\text{ kPa}$   $p_a = 90,000\text{ Pa}$   $p_w = 0.622 p_t W = 0.622 \times 90,000 \times 0.015 = 837.9\text{ Pa}$   $p_a = p_t - p_w = 90,000 - 837.9 = 89,162.1\text{ Pa}$   $v_a = \frac{R_a t_s}{p_a} = \frac{287 \times 303}{89,162.1} = 0.97\text{ m}^3/\text{kg}$

### or CLICK HERE

Solution: Perfect-Gas Equation:  $R T p = v p R T = v A t$  At 20 C, Table A-1, Saturation Pressure = 2.337 kPa = 2337 Pa. Specific volume of saturated vapor = 57.84 m<sup>3</sup> /kg  $T = 20\text{ C} + 273 = 293\text{ K}$   $( ) ( )$  Pa 2337 K 293 J/kg.K 462 =  $v$  /kg m 57.923 3 =  $v$  ( ) 100% 57.84 57.84 57.923 Deviation = Deviation = 0.1435 % 2-5.

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