

## Density Laboratory Gizmo Answers

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### Density Laboratory Gizmo Answers

Gizmo Density Lab Answers To calculate an object's density, divide its mass by its volume. If mass is measured in grams and volume in cubic centimeters, the unit of density is grams per cubic centimeter ( $\text{g/cm}^3$ ). Calculate the density of each object, and record the answers in the last column of your data table.

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If the mass is larger, the object will sink. If the volume is larger, the object will float. You could also find the density of the object by dividing mass by volume. If the density is greater than  $1 \text{ g/cm}^3$ , the object will sink. If the density is less than  $1 \text{ g/cm}^3$ , the object will float.

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To calculate an object's density, divide its mass by its volume. If mass is measured in grams and volume in cubic centimeters, the unit of density is grams per cubic centimeter (g/cm<sup>3</sup>). Calculate the density of each object, and record the answers in the last column of your data table. Label this column "Density (g/cm<sup>3</sup>).".

### **Student Exploration: Density Laboratory**

Density Laboratory. Launch Gizmo. With a scale to measure mass, a graduated cylinder to measure volume, and a large beaker of liquid to observe flotation, the relationship between mass, volume, density, and flotation can be investigated. The density of the liquid in the beaker can be adjusted, and a variety of objects can be studied during the investigation.

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### **Gizmo Density Lab Answers**

Density Answer Key Vocabulary: density, mass, matter, volume Prior Knowledge Questions (Do these BEFORE using the Gizmo.) [Note: The purpose of these questions is to activate prior knowledge and get students thinking, Students are not expected to know the answers to the Prior Knowledge Questions,] 1. List three objects that you think would sink in water, and three objects you think would float.

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