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chemistry chapter 10 study guide. Flashcard maker : Lily Taylor. Gram-atomic mass. the molar mass of an element expressed in grams per mole (g/mol) Gram-formula mass. The mass of a mole of formula units in a ionic compound bond. Mole. the simplest structural unit of an element or compound.

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Sample Study Sheet 10.1: Basic Equation Stoichiometry - Converting Mass of One Compound in a Reaction to Mass of Another Sample Study Sheet 10.2: Limiting Reactant Problems Sample Study Sheet 10.3: Equation Stoichiometry Problems To get a review of the most important topics in the chapter, fill in the blanks in the Key Ideas section.

Chapter 10 Chemical Calculations and Chemical Equations

7.16.05 g/mol. 8.1.20 1024. Section 10.4 Empirical and Molecular Formulas. 1. The percent composition of a compound is the percent by mass of each of the elements in a compound. 2. Divide the mass of each element in the sample by the mass of the sample. Then multiply each quotient by 100. 3.c.

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Organic Chemistry Chapter 10. Organolithium reagents. addition to formaldehyde. addition to formaldehyde mechanism. addition to aldehydes. $R-X$ and $2Li = Li-X$ and $R-Li$. formation of primary alcohols. Run in ether. formation of secondary alcohols.

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